Glycosyl Transfer to Nitrogen via Cycloaddition

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ABSTRACT

This letter describes the reduction to practice of a novel concept for functionalization of the anomeric carbon of carbohydrates with a nitrogen substituent. Thus, bisheterodienes with a thiono sulfur terminus and a sulfonylimine terminus are shown to undergo cycloaddition smoothly and stereoselectively to three different glycals.

Glycoproteins are a class of molecules under intense study by glycobiologists.1,2 As a consequence of this interest, the organic chemistry required for the synthesis of glycosylated amino acids and peptides has been a rapidly developing area of research. The synthesis of N-linked glycoside derivatives is almost universally accomplished by derivatizing a glyco $syl-NH₂$ species.³ Two recent examples of "the state-of-theart" from the Danishefsky lab and the Meldal/Bock/Paulsen team describe polymer-bound methods. The former uses a resin-attached trisaccharide-NH2 coupling to tri- and pentapeptides in solution.4 The latter uses solution-phase glycosyl donors reacting with peptides attached to a PEG-based resin.⁵ The one exception to this generalization is the modfied Ritter reaction described by Fraser-Reid.⁶ We wish to describe a second exception: our cycloaddition approach depicted in eq 1 where $X = S$ and $Y = N$ which delivers functionalized N to C-1 of glycals.

Our earlier work, where $X = S$ and $Y = O$, used the computed gap between the LUMO of the heterodiene and the HOMO of the glycal as a predictor of successful cycloaddition.7 After completing a computational survey of potential substitutents for imine nitrogens, we settled on the sulfonylimine function as being suitable for production of a low-lying LUMO for our heterodiene. As the model case, we chose methyl acetoacetate *N*-phenylsulfonylimine **1** (shown as the predominant enamine tautomer), easily

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prepared from methyl acetoacetate by heating with benzenesulfonamide. Phthalimidosulfenylation of **1** smoothly afforded derivative **2**, the key precursor to the transient thiono sulfonylimine **3**. When **2** was treated with pyridine in the presence of tribenzyl glucal **4**, tribenzyl galactal **5**, and tribenzyl allal **6,** the cycloadducts **7**, **8**, and **9** were produced in very good yield (Scheme 1).

^a PhSO2NH2, TsOH, benzene, 95%. *^b* TMSCH2CH2SO2NH2, TsOH, benzene, 93%. *^c* PhthNSCl, **2a** >95%, **2b** 95%. *^d* Pyridine, CH_2Cl_2 .

The experiments were also repeated with the TMSethylsulfonyl N-protecting group developed by Weinreb,⁸ with similar results. The TMS-ethyl group can be cleanly removed with CsF to afford the NH species **8c**. Preliminary ozonolysis experiments with adducts **7a** and **8c** showed that the bicyclic species could be cleaved to produce the glycosylamide function illustrated in compounds **10** and **11** (Scheme 2).

As an alternate route to the key sulfonylimino thione **3**, we took advantage of the reversibility of the anthracene adduct **14** of thiono keto ester **13**. ⁹ Adduct **14** was converted to its oxime **15**. The oxime, when treated with toluenesulfinyl chloride, undergoes sulfinylation followed by rearrangement

to the sulfonylimine **16**. ¹⁰ Then, mild thermolysis of this modified anthracene adduct in the presence of glycals **4** and **5** liberates the transient thiono species **3c** ($R_2 = Et$) which is smoothly trapped in cycloaddition to afford **7a** and **8a** $(R_2 = Et)$ (Scheme 3).

^a Pyridine (1 equiv), anthracene, 90%. *^b* NH2OH, 85%. *^c* PhSOCl, Et₃N, Et₂O/CH₂Cl₂ (1:1), -10 ^oC to rt. 20 h, 40%. *d* Lutidine (0.2) equiv), CHCl₃ 60 °C.

To develop our cycloaddition for a direct transfer to a model peptide, we modified aspartic acid as shown in

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Scheme 4. Blocked aspartic acid derivative **17** was converted to keto ester **18** using Meldrum's acid as a nucleophilic

a (i) Isopropenyl chloroformate, DMAP, Et₃N; (ii) EtOH, heat, 92% for two steps. *^b* NH2OH. *^c* TsCN, Et3N, CCl4, 65% for two steps. *^d*PhthNSCl.

C-donor in a peptide coupling protocol following the work of Shiori.11 The ketone was then converted to its sulfonyl oxime **¹⁹** according to a sulfinylation-rearrangement procedure reported by Boger.12 Phthalimidosulfenylation of **20** followed by cycloaddition led smoothly to glycopeptide analogue **23**, thus validating our concept. In conclusion, we have described a unique method for introducing functionalized nitrogen to the anomeric carbon of carbohydrates.13,14

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(14) **Representative Procedures. Phthalimidosulfenylation of methyl 3-benzenesulfonylamino-2-butenoate.** To a solution of sulfonyl imine **1a** (prepared via refluxing a solution of methyl acetoacetate (1.2 equiv) and benzenesulfonylamide (1 equiv) in a convenient amount of benzene with a catalytic amount of *p*-toluenesulfonic acid followed by conventional workup) was added PhthN-S-Cl (1.2 equiv) in portions at 0° C during a period of 15 min. The reaction mixture was stirred at such temperature for an additional 20 min and allowed to warm to room temperature in 30 min. Cold *n*-pentane was added. A lot of white precipitate formed which was filtered and then washed with cold *n*-pentane to afford the desired **2a**. Yield of crude suitable for the following step: $>99\%$; ¹HNMR (CDCl₃) δ 12.53 (s, 1H), 7.95-7.57 (m, 9H), 3.80 (s, 3H), 2.89 (s, 3H). **Cycloaddition of methyl 2-phthalimidosulfenyl**-**3-benzenesulfonylamino-2-butenoate 2a** with tri-*O***-benzyl-**D-glucal (4). To a solution of the phthalimidosulfenyl imine 2a (1.2 equiv) and tri-O-benzyl-D-glucal (4) in CHCl₃ was added a catalytic amount of 2,6-lutidine (2 mol %). The resulting solution was stirred at room temperature until the reaction was complete as monitored by TLC. The solution was dissolved in dichloromethane and washed with saturated ammonium chloride and brine and dried over Na2SO4. The organic solvent was removed under reduced pressure. The crude materials were purified by a silica gel column (ethyl acetate/petroleum ether) to give the desired product **7a**. When phthalimide residues are not cleanly separated, a 20% NaOH wash was used to extract the phthalimide after flash chromatography. Yield of **7a**: 82% (140 mg, 0.35 mmol); FTIR (neat) 1715.3, 1585.4, 1448.2, 1357.6, 1251.5, 1169.1; ¹HNMR (CDCl₃) δ 7.96 (d, 2H), 7.52 (t, 2H), 7.40-1357.6, 1251.5, 1169.1; ¹HNMR (CDCl₃) *δ* 7.96 (d, 2H), 7.52 (t, 2H), 7.40-
7 16 (m) 6 33 (d, J = 7 2, 1H), 4 82–4 57 (m), 3 71 (s, 3H), 3 52–3 3 7.16 (m,), 6.33 (d, *J* = 7.2, 1H), 4.82–4.57 (m,), 3.71 (s, 3H), 3.52–3.3
(m 4H) 2.52 (s 3H)^{, 13}CNMR (CDCl3) δ 165 2, 148 1, 139 2, 137 9, 137 6 (m, 4H), 2.52 (s, 3H); 13CNMR (CDCl3) *δ* 165.2, 148.1, 139.2, 137.9, 137.6, 133.1, 128.6, 128.2, 128.0, 127.7, 127.6, 127.5, 117.4, 87.9, 79.0, 77.9, 76.1, 74.9, 73.3, 72.0, 67.5, 52.1, 47.9, 21.4. Anal. Calcd for C₃₈H₃₉O₈-NS2: C, 65.03; H, 5.56; N, 2.00. Found: C, 64.67; H, 5.93; N, 2.12.

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⁽¹³⁾ All cycloadducts had molecular ions (either ESMS or CI/MS) and proton and carbon NMR data consistent with their structures. Adducts **7a**, **8b**, **10,** and **11** have confirming combustion analyses and the structure of **7a** has been confirmed by X-ray crystallography.